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Paul et al. are the metalloids [438] ~ Rochow [444] who later wrote his 1966 monograph The metalloids, [445] commented that, 'In some respects selenium acts like a metalloid and tellurium certainly does'. 11, nos 11-12, pp. 1377-82, doi:10.1016/j.interm.2003.09.008 Miller GJ, Lee C & Choe W 2002, 'Structure and Bonding Around (ZnCl)2 in (Gymer), D Naumann & L Wesermann (eds), Inorganic chemistry highlights, Wiley-VCH, Weinheim, pp. 21-53, ISBN 3-527-30265-4 Millot F, Rifflet JC, Sarou-Kanian V & Wille G 2002, 'High-Temperature Properties of Liquid Boron from Intersect Techniques', *International Journal of Thermophysics*, vol. 11, pp. 251-74, doi:10.1002/SZT.1520-670X(1998)11:233:0.CO;2-Q NIST (National Institute of Standards and Technology) 2010, Ground Levels and Ionization Energies for Neutral Atoms, by WC Martin, A Musgrave, S Kotchigova & JE Sansonetti, viewed 8 February 2013 National Research Council 1984, The Competitive Status of the U.S. Electronics Industry: A Study of the Influences of Technology in Defining International Industrial Competitive Advantage, National Academy Press, Washington, DC, ISBN 0-309-03397-7 New Scientist 1975, 'Chemistry on the Islands of Stability', 11 Sep, p. 574, ISSN 0132-1233 New Scientist 2014, 'Colour-changing metal to yield thin, flexible displays', vol. 111, pp. 11604-1 + 1604-5, doi:10.1103/PhysRevLett.111.16044 Herold A 2006, 'An Arrangement of the Chemical Elements in Several Classes Inside the Periodic Table according to their Common Properties', *Comptes Rendus Académie des Sciences Paris*, vol. 27, pp. 115-18, doi:10.1016/j.crd.2006.01.001. ... suggested classifying some elements as metalloids "emphasizes that properties change gradually rather than abruptly as one moves across the table periodic table" [81] Some periodic tables distinguish elements that are metalloids and display no formal dividing line between the metalloids and nonmetals, 87, 1999, p. 501. Craig & Maher 2003, p. 391. Stuedel 1977, p. 240. "... considerable overlap must exist, to form intermolecular, many-center... 1999, p. 498. Wells 1984, p. 715. Wiberg 2001, p. 588. Mellor 1964a, p. 30. Wiberg 2001, p. 589. Greenwood & Earnshaw 2002, pp. 765-66. Baugnall 1966, pp. 134-51; Greenwood & Earnshaw 2002, p. 786. Drey & O'Regan 1994, pp. 1-2. Hill & Holman 2000, p. 124. Chang 2002, p. 314. Clark 1950, pp. 1-2; Clark 1950, pp. 588; Warren & Geballe 1981, p. 205; Housecroft & Sharpe 2008, p. 384; IUPAC 2006, 'rhomboidal graphite entry' *Mingos* 1998, p. 171. Wiberg 2001, p. 781. Charlier, Gonze & Michenaud 1994, p. a b c Atkins et al. Mendeleev 1871 table 1869. ... predictions Discovery of elements Naming and etymology for people for places continents (in East Asia) Systematic element names Sets of elements By periodic table structure Groups (1-18) (alkali metals) 2 (alkaline earth metals) 3 4 5 6 7 8 9 10 11 12 13 14 15 (pnictogens) 16 (chalcogens) 17 (halogens) 18 (noble gases) Periods (1-7, ...) 1 2 3 4 5 6 7 8 Blocks (s, p, d, f, ...) Atomic orbitals Aufbau principle By metallic classification Metals alkali alkane earth transition post-transition lanthanide actinide (superactinide) Metalloids dividing metals and nonmetals Nonmetals unclassified nonmetal halogen noble gas By other characteristics Coinage metals Platinum-group metals Precious metals Refractory metals Heavy metals Light metals Native metals Noble metals Main-group elements Rare-earth elements Transuranium, transplutonium elements Major (minor) and trace actinides Elements List of chemical elements by abundance (in human body) by atomic properties by isotope stability by annual production by symbol Properties of elements (1-108) The periodic table (1-108) Polymeric [274] [266] [267] [268] [269] [270] [271] [272] [273] [274] [275] [276] [277] [278] [279] [280] [281] [282] [283] [284] [285] [286] [287] [288] [289] [290] [291] [292] [293] [294] [295] [296] [297] [298] [299] [300] [301] [302] [303] [304] [305] [306] [307] [308] [309] [310] [311] [312] [313] 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